

Certificate of Analysis

Standard Reference Material 6g

Cast Iron

| ANALYST | C | | Mn | P | S | Si | Cu | Ni | Cr | V | Mo | Ti | As | N |
|---------|-------|----------|-------------------------|---|----------------------|-------------------------------|--------------------|-------------|--------------------|--------------------|-------------|--------------------|-------------|--------------------------|
| | Total | Graphite | Peroxydisulfate-arsenic | Photometric | Combustion-titration | HClO ₄ dehydration | Photometric | Photometric | | | Photometric | Photometric | Photometric | Distillation-Photometric |
| 1 | 2.82 | 2.01 | 1.07 | { 0.555 ^a .551 ^b } | 0.126 ^c | 1.06 ^d | 0.499 ^e | 0.136 | 0.360 ^f | 0.059 ^g | 0.038 | 0.052 ^h | 0.043 | 0.006 ⁱ |
| 2 | 2.85 | 2.01 | 1.05 | .561 | .122 | 1.06 | .502 | .136 | .375 | .050 | .033 | .060 | .041 | .004 |
| 3 | 2.83 | 1.99 | 1.06 | .552 ^a | .126 | 1.05 ^d | .498 ^j | .134 | .376 ^k | .058 ^l | .037 | .060 ^m | --- | .006 ⁿ |
| 4 | 2.86 | 2.04 | 1.03 | .562 | .122 | 1.07 | .51 ^o | .134 | .372 ^p | --- | --- | --- | --- | --- |
| 5 | 2.91 | 2.00 | 1.04 | .562 ^q | --- | 1.01 ^r | --- | --- | .369 ^s | --- | .032 | .065 ^t | --- | .005 ⁿ |
| Average | 2.85 | 2.01 | 1.05 | 0.557 | 0.124 | 1.05 | 0.502 | 0.135 | 0.370 | 0.056 | 0.035 | 0.059 | 0.042 | 0.005 |

^a Molybdenum-blue photometric method.

^b Mg₂P₂O₇ gravimetric method.

^c i-g sample burned in oxygen at 1425 °C and sulfur dioxide absorbed in starch-iodide solution. Iodine liberated from iodide by titration, during the combustion, with standard KIO₃ solution.

^d Double dehydration with intervening filtration.

^e Atomic absorption spectrometry.

^f Coulometric method.

^g Activation analysis.

^h Polarographic method.

ⁱ Distillation-indophenol photometric method.

^j Copper-ammonia complex photometric method.

^k Peroxydisulfate oxidation-Fe(NH₄)₂(SO₄)₂ titration.

^l Nitric acid oxidation-Fe(NH₄)₂(SO₄)₂ titration.

^m Vanadium separated with Na₂CO₃.

ⁿ Distillation-photometric with Nessler's reagent.

^o Diethyldithiocarbamate photometric method.

^p Diphenylcarbazide photometric method.

^q Alkali-molybdate method.

^r Double dehydration with H₂SO₄.

^s Bicarbonate hydrolysis-Fe(NH₄)₂(SO₄)₂-KMnO₄ titration.

^t Cupferron separation-H₂O₂ photometric method.

The material for the preparation of this standard was furnished by the American Cast Iron Pipe Company, Birmingham, Alabama.

The overall direction and coordination of the technical measurements leading to certification were performed under the chairmanship of O. Menis and J. I. Shultz.

The technical and support aspects involved in the preparation, certification, and issuance of this standard reference material were coordinated through the Office of Standard Reference Materials by R. E. Michaelis.

Washington, D. C. 20234
 November 9, 1970
 (Replaces Prov. Cert. 10/16/67)

J. Paul Cali, Acting Chief
 Office of Standard Reference Materials

(over)

List of Analysts

1. J. R. Baldwin, B. B. Bendigo, R. W. Burke, D. A. Becker, P. D. LaFleur, G. Marinenko, E. J. Maienthal, E. R. Deardorff, T. C. Rains, T. A. Rush, and S. A. Wicks. Analytical Chemistry Division, Institute for Materials Research, National Bureau of Standards.
2. R. E. Deas, R. N. Smith and J. B. Hobby, American Cast Iron Pipe Company, Birmingham, Alabama.
3. C. R. Vinyard and F. Wojnarowski, Jones and Laughlin Steel Corporation, Cleveland Works, Cleveland, Ohio.
4. W. B. Sobers, Rex Chainbelt Incorporated, Milwaukee, Wisconsin.
5. L. E. Chalker, The Youngstown Sheet and Tube Company, Youngstown, Ohio.