



## METHODS USED

Element	Line number	Analytical methods
C	1, 3, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16 2 4	Combustion - infrared absorption Coulometric titration Non-aqueous titration after absorption in organic solvent
Si	1, 8, 10, 13, 3, 4, 5, 6, 7, 9, 11, 12, 15 14	ICP - OES Gravimetry, dehydration with perchloric acid Gravimetry, dehydration with hydrochloric acid
Mn	1, 3, 4, 6, 7, 8, 9, 10, 11, 14, 17 2, 5, 13, 15 12, 16	ICP - OES FAAS MAS, periodate oxidation
P	1, 3, 4, 8, 12, 14 2, 9, 13 5 6, 7, 10, 15 11	ICP - OES MAS, molybdenum blue, without extraction Acidimetric titration of ammonium phosphomolybdate MAS, phosphovanadomolybdate, extraction MAS, molybdenum blue, extraction
S	1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 12, 13, 14, 15, 16 11	Combustion - infrared absorption Gravimetry as BaSO <sub>4</sub> without separation
Cr	1, 5, 6, 7, 8, 9, 10, 12 2 3, 4, 13, 14, 15 11	Titration with Fe (II), oxidation with persulphate FAAS ICP - OES Titration with Fe (II), oxidation with perchloric acid
Mo	3, 4, 6, 7, 8, 9, 13, 17 5, 11, 15, 16 10, 12, 14	ICP - OES FAAS MAS, thiocyanate in presence of Sn (II), extraction
Ni	1, 5, 6, 10, 13 2, 3, 4, 8, 9, 11, 12, 14, 15, 17 7	FAAS ICP - OES MAS, dimethylglyoxime, extraction
Cu	2, 3, 5, 6, 8, 10, 12, 13, 16 4, 7, 9, 11, 14, 15, 17	ICP - OES FAAS
N	1 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16	Acidimetric titration after distillation, visual end point Thermal conductivity, decomposition in graphite crucible
V	3, 4, 9, 12, 13, 14, 15, 20 5, 6, 8, 11, 17, 18, 19 7, 10, 16	Titration with Fe (II), oxidation with Mn (VII) ICP - OES FAAS
W	1 2, 4, 8, 10, 11, 14, 15 3, 5, 6, 9, 12, 13 7	MAS, thiocyanate formed in a strongly acid reducing medium ICP - OES ICP - MS NAA
Al <sub>(tot)</sub>	1 2, 3, 5, 8, 10 4, 6, 7, 9, 11	ETAAS FAAS ICP - OES
O	1, 2, 3, 4, 5, 6, 7, 8, 9, 11, 12, 13, 14, 15, 16 10	Infrared absorption measurement reduction fusion under helium Thermal conductivity, reduction fusion under nitrogen
As	1, 2 3	ICP - MS NAA
Co	1, 3, 4 2	ICP - MS ICP - OES
Pb	1, 2, 3	ICP - MS
Sn	1, 2, 3 4	ICP - MS ICP - OES
Ti	1 2,3	ICP - OES ICP - MS
Sb	1, 2, 3	ICP - MS

**Abbreviations:**

FAAS	Flame Atomic Absorption Spectrometry
ICP - OES	Inductively Coupled Plasma - Optical Emission Spectrometry
ICP - MS	Inductively Coupled Plasma - Mass Spectrometry
ETAAS	Electrothermal Atomic Absorption Spectrometry
MAS	Molecular Absorption Spectrometry
NAA	Neutron Activation Analysis

## **DESCRIPTION OF THE SAMPLE**

The ECRM 274 -1 is available in the form of milling chips in bottles containing 100 g. It is also available as 38 mm diameter discs 25 mm thick. The chips were passed through a 2000 µm aperture sieve and further sieving was carried out to exclude chips passing through a 250 µm aperture sieve.

## **INTENDED USE & STABILITY**

The chip sample, ECRM 274-1, is intended for the verification of analytical methods, such as those used by the participating laboratories, for the calibration of analytical instruments in cases where the calibration with primary substances (pure stoichiometric metals or compounds) is not possible and for establishing values for secondary reference materials.

It will remain stable provided that the bottle remains sealed and is stored in a cool, dry atmosphere. When the bottle has been opened the lid should be secured immediately after use. If the contents should become discoloured (eg oxidised) due to atmospheric contamination they should be discarded.

The solid (disc) sample, ECRM 274-1, is intended for establishing and checking the calibration of Optical Emission and X-ray Spectrometers for the analysis of samples of similar materials. The "as received" working surface of the sample should be finished before use to remove any protective coating. It will remain stable provided that it is not subjected to excessive heat (eg, during preparation of the working surface).

## **TRACEABILITY**

The traceability of this ECRM is ensured by the use of either stoichiometric analytical techniques or methods which are calibrated against pure metals or stoichiometric compounds.

## **FURTHER INFORMATION**

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A handwritten signature in blue ink that reads "Rein Vainik".

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